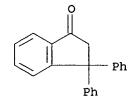
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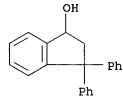
ΑN 1989:113878 CAPLUS 110:113878 DN ΤI Reactions of carboxylic acids with phosphonium anhydrides Hendrickson, James B.; Hussoin, M. Sajjat ΑU Edison Chem. Lab., Brandeis Univ., Waltham, MA, 02254, USA CS SO J. Org. Chem. (1989), 54(5), 1144-9 CODEN: JOCEAH; ISSN: 0022-3263 DTJournal LA English os CASREACT 110:113878; CJACS AΒ General considerations are outlined for a reagent to ext. oxygen from org. mols. by an equiv. of dehydration. The reagent (Ph3P+)2O 2(OTf)- (OTf = triflate) was created for the purpose and subjected to a preliminary study. The reagent convert carboxylic acids readily and rapidly to anhydrides, esters, amides, amidines, benzimidazoles, and cyclic aryl ketones in good yields. Thus, treatment of 4-MeC6H4CO2H with Ph3PO in the presence of triflic anhydride and Et3N gave 93% p-toluic anhydride. IT 55010-17-8P RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. of) RN 55010-17-8 CAPLUS CN 1H-Inden-1-one, 2,3-dihydro-3,3-diphenyl- (9CI) (CA INDEX NAME)



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1971:75859 CAPLUS AΝ DN 74:75859 TI Thermolysis of substituted indenes. Sigmatropic phenyl and hydrogen migrations ΑU Miller, Larry Lee; Boyer, Rodney F. Dep. Chem., Colorado State Univ., Fort Collins, Colo., USA CS J. Amer. Chem. Soc. (1971), 93(3), 650-6 SO CODEN: JACSAT DΤ Journal LΑ English 1,1,3-Triphenylindene, 1,1-diphenylindene, 1-methyl-1-phenylindene, and AB 1,3-diphenylindene rearrange at 250-300.degree. via a 1,2-phenyl migration. The resp. products formed are 1,2,3-triphenylindene, 2,3-diphenylindene, 3-methyl-2-phenylindene, and 2,3-diphenylindene. These reactions in Ph2O are kinetically first order. The rate const. for 1,1,3-triphenylindene rearrangement is unaffected by added acid, base, or free-radical scavengers. .DELTA.S.noteq. for this phenyl migration is -25 entropy units. Solvation of the transition state for rearrangement accounts for a portion of this very neg. value as is indicated by the relative rates of rearrangement in solvent Decalin (2.45), Ph2O (8.34), .omicron.-cresol (8.8), and HCONMe2 (16.5). In contrast, H rearrangement from the 1 to the 2 position of 1-phenylindene shows no solvent effect and .DELTA.S.noteq. -2.3 entropy units. Studies of H (D) rearrangement in 1-deuterioindene, 1-phenylindene, and 1,3-diphenyl-1-deuterioindene at 150.degree. allow estn. of Ph substituent effects on sigmatropic H rearrangement. A 1-Ph accelerates migration by about 130 and 3-Ph by 6. Accelerative substituent effects on Ph migration are similar: 1-Ph (50), 3-Ph (5), 1-Me (8). These results are interpreted in terms of the transition state connecting reactant indene with an isoindene intermediate. The data reveal a migratory aptitude series H > Ph > Mewhich is detd. by the more effective bridging capabilities of H compared to C. IT 31366-71-9P RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. of) RN 31366-71-9 CAPLUS 1-Indanol, 3,3-diphenyl- (6CI, 8CI) (CA INDEX NAME) CN



RN 18636-52-7 CAPLUS CN 1H-Indene, 1,1-diphenyl- (9CI) (CA INDEX NAME)

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Beilstein Reg. No. (BRN): 4924895 Beilstein Molecular Formula (MF): C21 H17 N O . Cl H

Lin. Struct. Formula (LSF): C21H17NO*HCl

Chemical Name (CN): 3,3-diphenyl-indan-1-one oxime;

hydrochloride

3,3-Diphenyl-indan-1-on-oxim; Hydrochlorid

Beilstein Reference (SO): 2-07-00-00496

Component Data:

Reg. No. (CBRN)	1	Component Molec. Formula (CMF)	i	Weight (FW)	i 	Lawson Number
	i	C21 H17 N O		299.37 36.46	1	

CM 1

CBRN 3373716 CMF C21 H17 N O

CM 2

CBRN 1098214 CMF Cl H

Ring System Data:

3373716 Component BRN (CBRN): Number of Rings (CNR): Ring Systems (CNRS): Diff. Ring Systems (CNDRS): 2 Ring Heteros (CNRH): Acyclic Heteros (CNAH):

Beilstein Ri (BRIX)	ng	Index		Ring (RF)	System	Formula	 -	BRIX Count
9.2.5-0.0-3. 6.1.0-0.0-3.			•	C9 C6		- 1 ,	1	1 2

1098214 Component BRN (CBRN): Number of Rings (CNR): Acyclic Heteros (CNAH):

Field Availability:

Code	Name	Occur. (OCC)
MF LSF CN FW SO LN	Molecular Formula Linearized Structure Formula Chemical Name Formula Weight Beilstein Citation Lawson Number	1 1 2 2 1
SF MP	Stereo Family Melting Point	1 1

d mp

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Melting Point: Value | Ref. | Note (MP) | | (Cel) - 1 175.00 |1 | 1

1. Gagnon, Ann.Chim.(Paris), <10> 12 <1929>, 315, CODEN: ANCPAC

Notes(s):

1. Handbook Data